

# Preparative LC for Hg isotope ratio measurements of Hg species in fish by MC-ICPMS

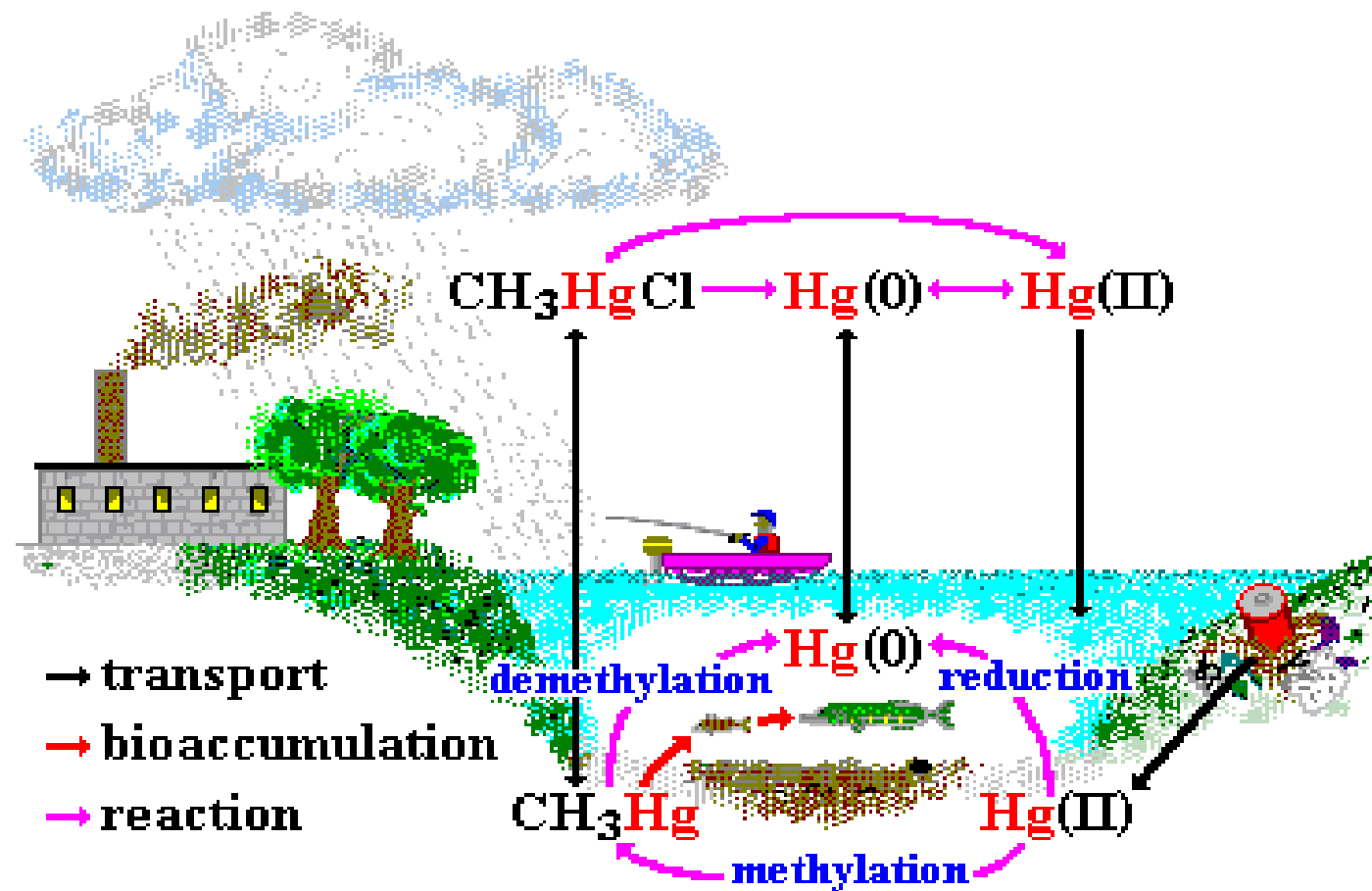


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Science  
for a safer world



# Mercury cycle in the biosphere



# Background



- "Minamata Convention on Mercury" A global and legally binding convention adopted at the UNEP Diplomatic Conference held in October 2013 in Japan.
- A Global Mercury Observation System was set up to monitor levels around the world.
- Hg isotopic signature information of the species can provide extra information on biotransformation and sources.

The project -Joint Research Project of the European Metrology Research Programme (ENV51 – MeTra)  
Work package 3 : Traceability for mercury isotopic measurements

- <http://projects.lne.eu/JRP/MeTra/project-overview/index.asp>



# History of LGC's contribution to accurate mercury speciation measurement



In 1997 Chris Harrington et al published a paper  
“Problems Encountered During the Development of a  
Method for the Speciation of Mercury and Methylmercury  
by High Performance Liquid Chromatography Coupled to  
Inductively Coupled Plasma Mass Spectrometry”  
*J. Anal. At. Spectrom.* 12, 1997, 1053-1056

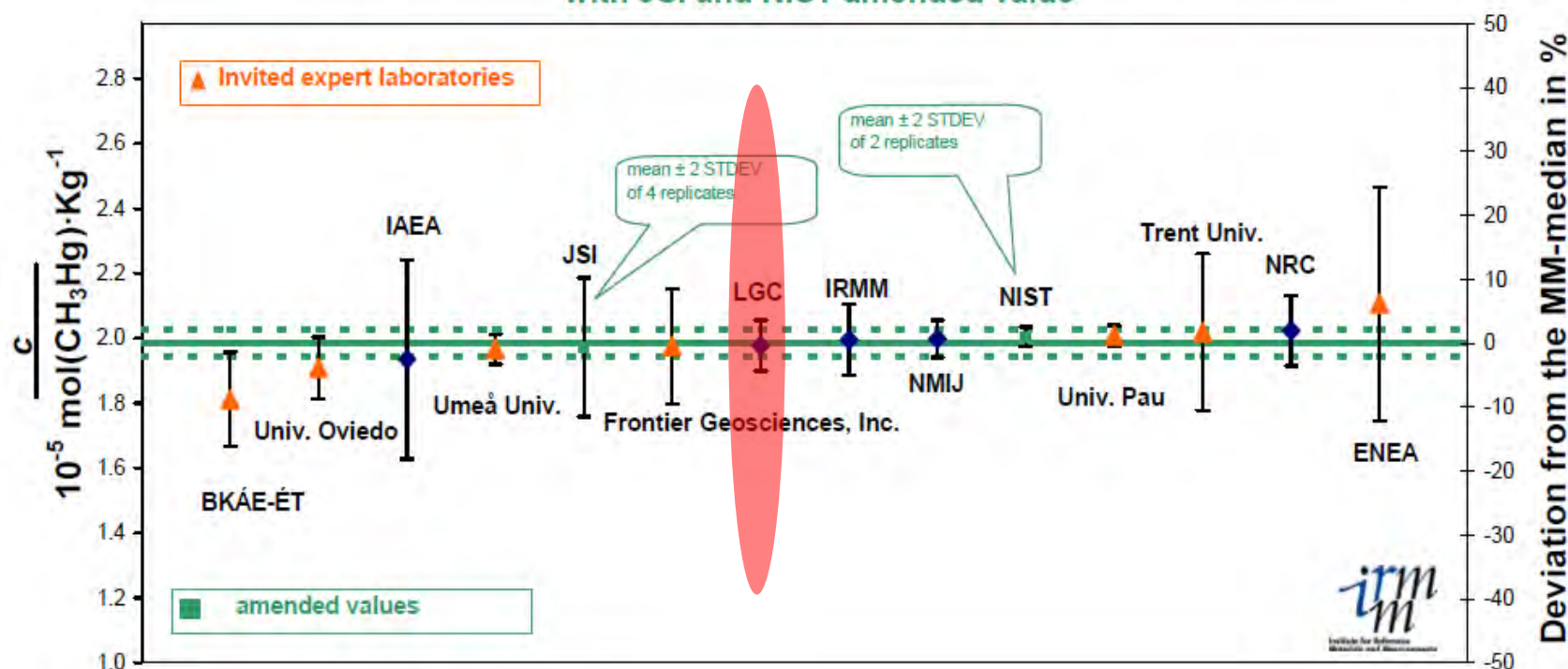
Solution - 2-mercaptoethanol was used to overcome poor  
peak shape and LC tailing problems.

# 2003 - accurate quantitation of Methylmercury in tuna (defatted fish)



## CCQM-P39: methylmercury in tuna fish

Mixture Model-median:  $1.970 \pm 0.042 \cdot 10^{-5} \text{ mol (CH}_3\text{Hg)·Kg}^{-1}$  ;  $[\mu \pm \sigma \cdot t_s / \text{sqrt}(n)]$   
with JSI and NIST amended value



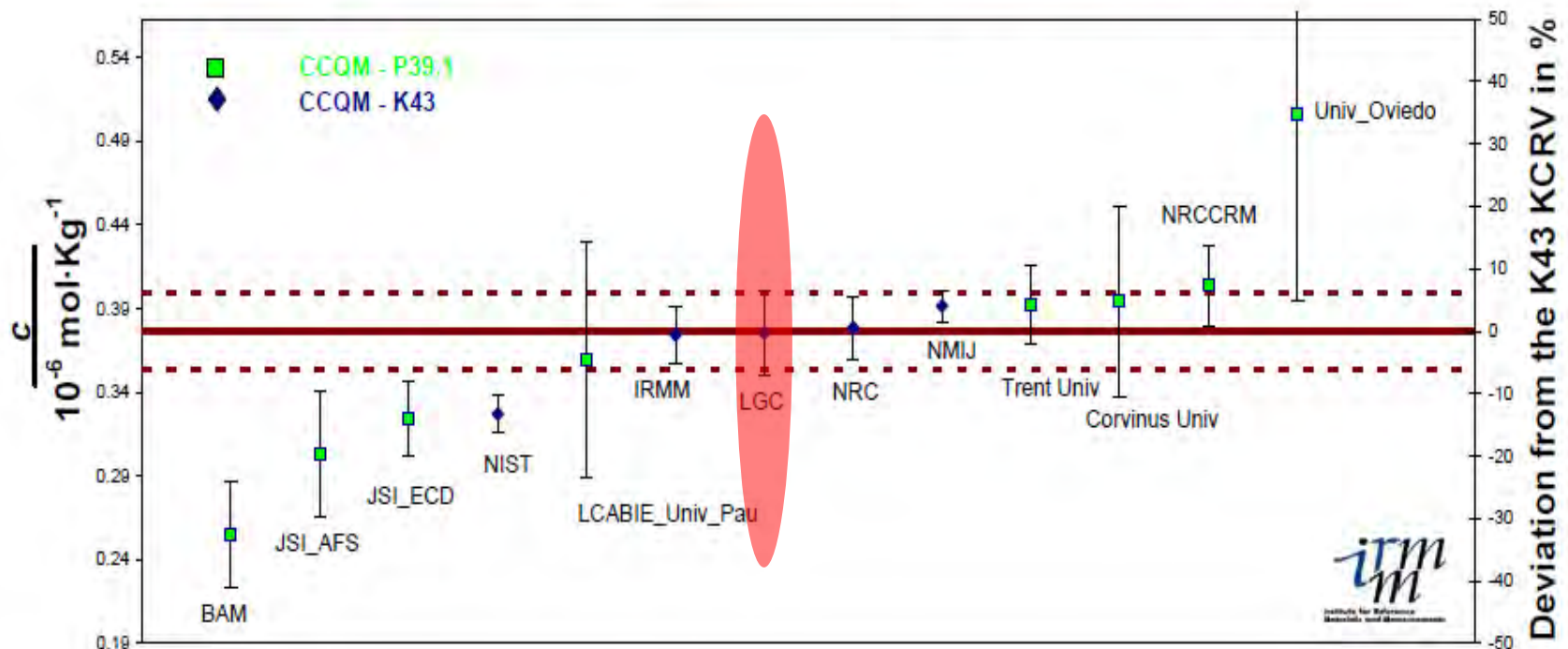
[http://www.bipm.org/utils/common/pdf/final\\_reports/QM/P39/CCQM-P39.pdf](http://www.bipm.org/utils/common/pdf/final_reports/QM/P39/CCQM-P39.pdf)

# 2005- accurate quantitation of Methylmercury in salmon (oily fish)



## CCQM-K43 & P39.1: MeHg in salmon

Mixture Model-median from CCQM-K43:  $0.373 \pm 0.023 \cdot 10^{-6} \text{ mol} \cdot \text{Kg}^{-1}$  ;  $[\mu \pm \sigma \cdot t_s / \sqrt{n}]$



[http://www.bipm.org/utils/common/pdf/final\\_reports/QM/K43/CCQM-K43.pdf](http://www.bipm.org/utils/common/pdf/final_reports/QM/K43/CCQM-K43.pdf)

# Development of the procedure for Hg isotope ratio measurements of Hg species



## Steps in the analytical process

- Extraction.
- LC separation.
- Isolation and collection.
- Minimise dilution and concentrate extract.
- Measurement of isotope ratio using multi-collector inductively coupled plasma mass spectrometer (MC-ICP-MS) (20ng g<sup>-1</sup> minimum concentration)

# Fractionation of Hg isotopes



- Mercury has 7 naturally occurring isotopes
- In nature the maximum isotopic variability is in the order of 0.8%

Mass Number	Natural Abundance
196	0.15%
198	9.97%
199	16.87%
200	23.10%
201	13.18%
202	29.86%
204	6.87%



# Selection of sulfhydryl group complexing agent used during both extraction and LC.

## 2-mercaptoethanol

### Advantage

- Stable

### Disadvantage

- $\text{Hg}^{2+}$  eluted in tail of larger  $\text{CH}_3\text{Hg}$  with low organic modified
- Volatile
- Toxic (bad odour)

## Cysteine

### Advantage

- $\text{Hg}^{2+}$  eluted before larger  **$\text{CH}_3\text{Hg}$** .
- Non-volatile
- Non toxic

### Disadvantage

- Easily oxidises

# Extraction

## Challenges

- Quantitative extraction of all species.
- Possible transformation of the species ( $\text{CH}_3\text{-Hg} \leftrightarrow \text{Hg}^{2+}$ ).
- Need to minimise dilution.
- Loss of complexing ability by air oxidation of sulfhydryl group.

# Extraction

## Solutions

- Recoveries in good agreement with certified values of CRMs e.g. NIST SRM 1947
- Enzymatic hydrolysis of protein matrix.
- Moderate buffered pH of 7.5 and temperature 37°C.
- Released Hg species stabilised in solution with cysteine.
- Sonication decreases extraction time.
- Minimal sample dilution (1:12) incurred
- Confirmed minimal transformation of species by spiking experiments using labelled species ( $^{199}\text{Hg}^{2+}$  and methyl $^{202}\text{Hg}$ ).

# LC separation

## Challenges

- Require well resolved discrete peaks with minimal tailing.
- Minor species eluted before major species to avoid a collected fraction contain the tailing end of the peak
- Low organic modifier content of mobile phase will ease compatibility with ICP-MS.
- Isocratic is preferred to avoid complexity.
- Capacity for large injection volumes.

# LC separation

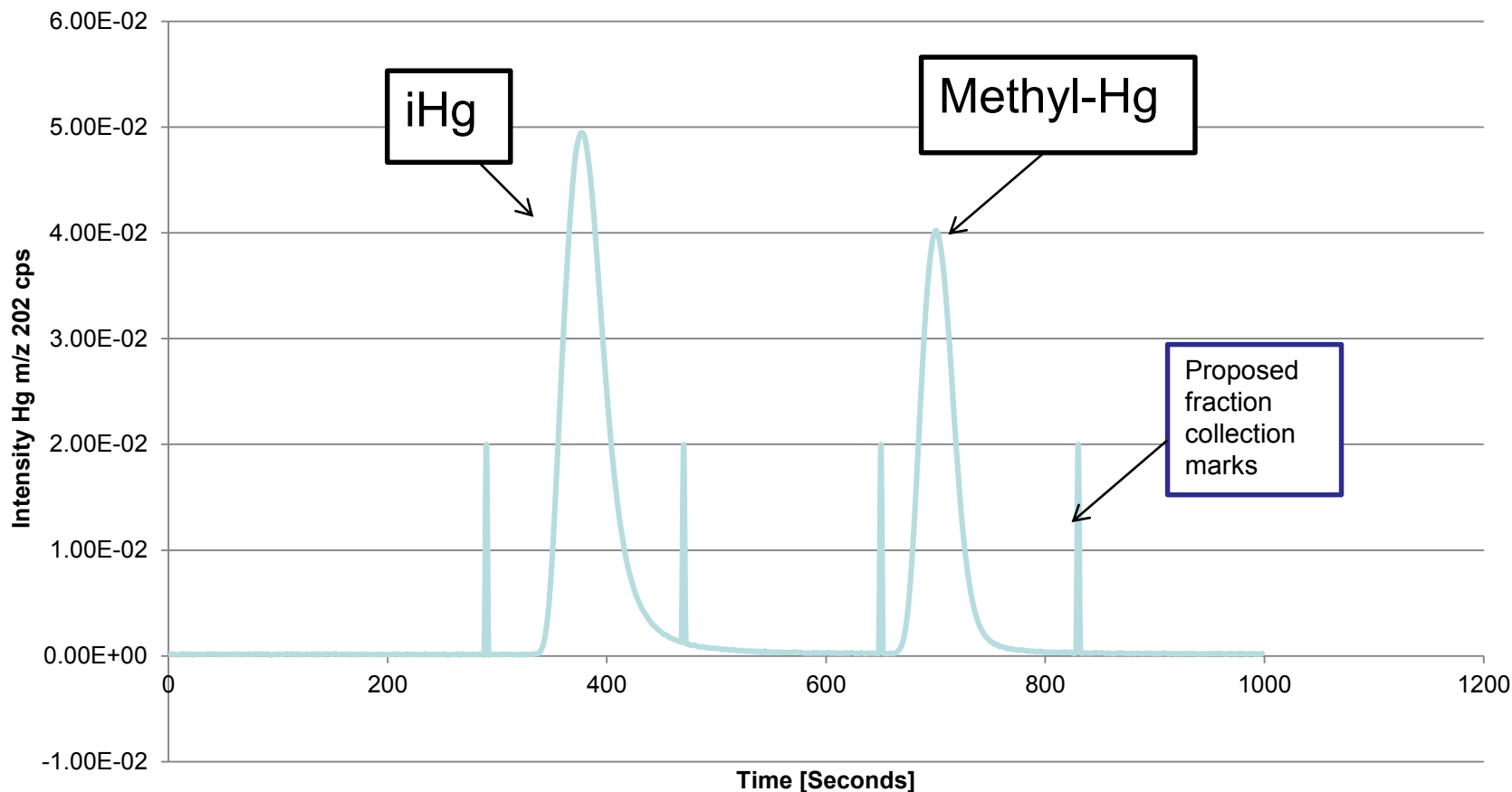
## Solutions

- Selection of a reverse phase stationary phase that does not collapse under high aqueous mobile phase.
- Wide bore enables increased loading.
- Selection of cysteine as a complexing reagent as  $\text{Hg}^{2+}$  elutes first with this reagent.
- Extracts fully compatible with chromatography enabling large injection volumes .
- 100% aqueous buffered phase mobile phase.

# Chromatogram using direct pneumatic nebulisation



Neptune 500 $\mu$ L injection of mixed stds 10ng/g



## Challenges

- Losses
- Sufficient concentration factor can be obtained.

## Solutions

- Multiple fractions can be collected and combined.
- Freeze drying of collected fraction.
- Use cold vapour [ $\text{Hg}^{2+} \rightarrow \text{Hg}^0$ ] increases signal intensity.

# Extract compatibility with cold vapour generation



## Challenges

- Methyl mercury will not react to form a volatile species
- Sulfhydryl complexing agent cysteine suppresses vapour generation.

## Solutions

- Microwave digestion with nitric acid (Methylmercury →  $\text{Hg}^{2+}$ )
- Cysteine → sulphate which has minimal complexing ability.



# Measurement of isotope ratio analysis



## Challenges

- Instrumental bias
- Fractionation from cold vapour generation.

## Solutions

- Thallium added on line to correct for instrument bias using known  $^{205}\text{Tl}/^{203}\text{Tl}$  isotope ratio of NIST SRM 997.
- Traceable NIST SRM instrumental fractionation caused by cold vapour generation process.

# Control of blank levels

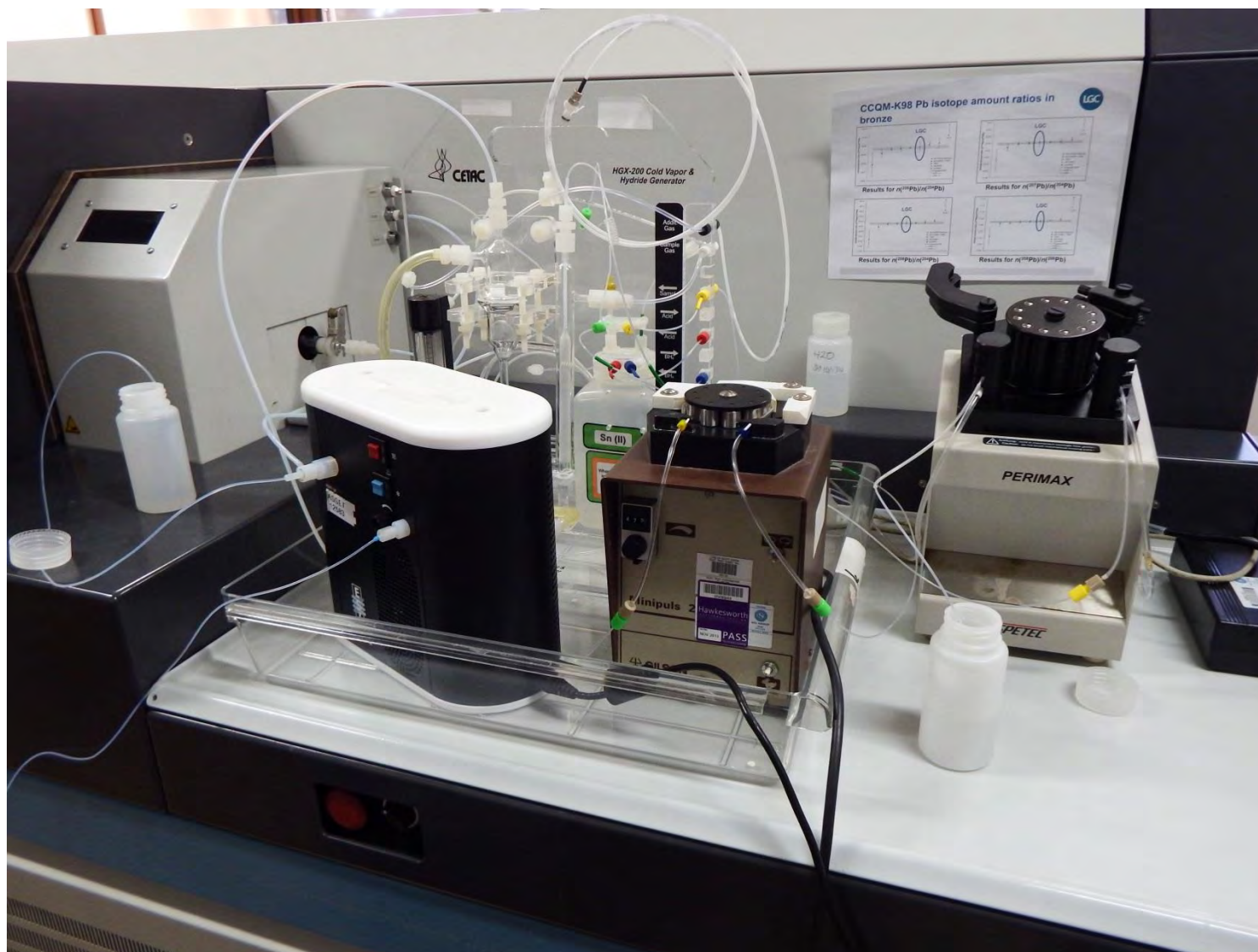
## Challenges

- Reagents
- Equipment
- Laboratory air.

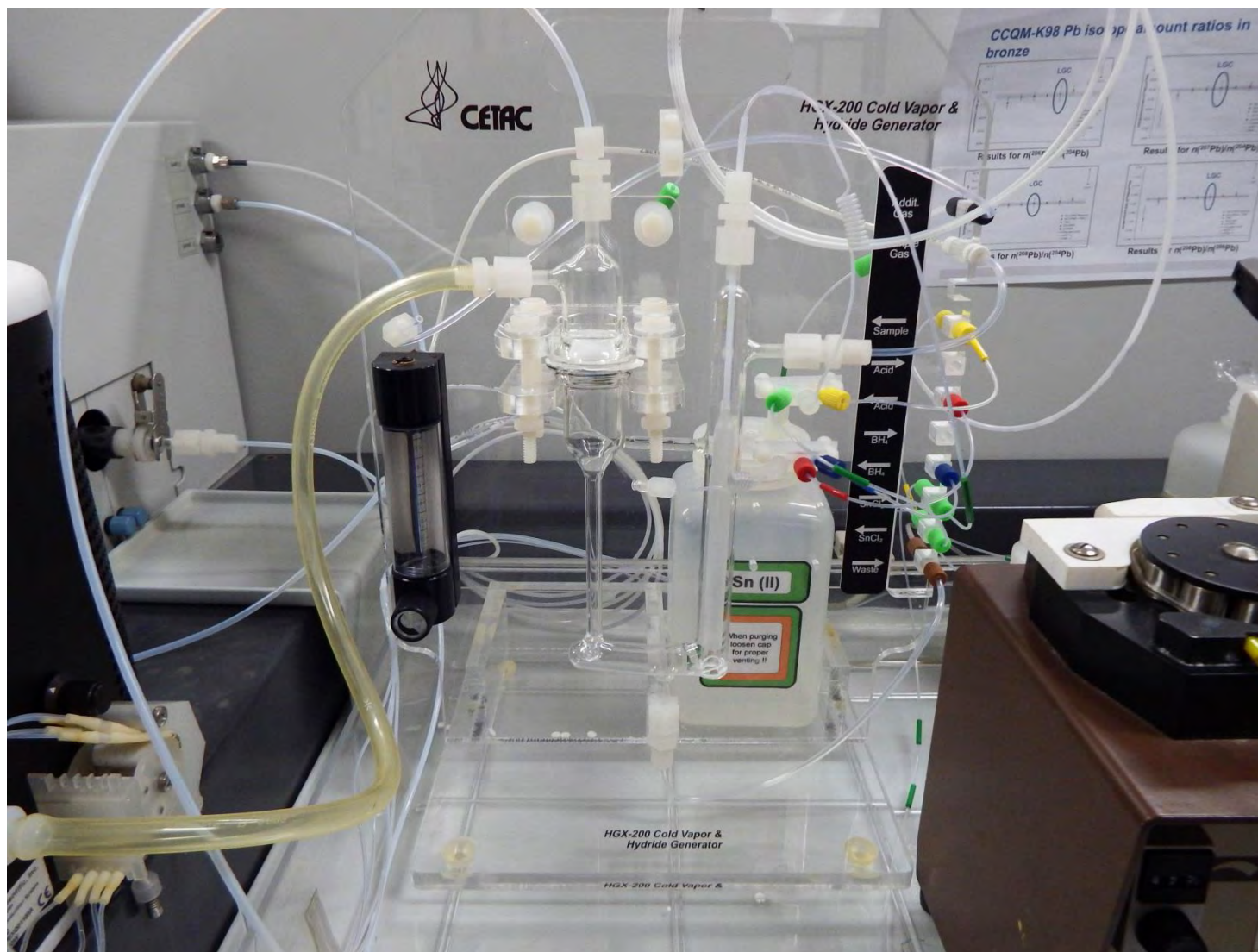
## Solutions

- pH meter probe “Hg free”.
- Enzymes derived from microorganism grown in culture media.
- Minimal exposure to lab air.

# Setup of MC-ICP-MS with cold vapour



# Setup of MC-ICP-MS with cold vapour



# Results of Hg isotope ratio measurements for fish tissues



There is no fish tissue reference material certified for the isotopic composition of mercury, either bulk or species-specific. Below are  $\delta$ -values for Hg isotope ratios for MeHg in fish tissues reference materials certified for Hg concentration ( $\text{‰} \pm$  standard deviation at  $2\sigma$  level).

	$^{199}\text{Hg}/^{198}\text{Hg}$	$^{200}\text{Hg}/^{198}\text{Hg}$	$^{201}\text{Hg}/^{198}\text{Hg}$	$^{202}\text{Hg}/^{198}\text{Hg}$
BCR 463	$1.9 \pm 0.2$	$0.3 \pm 0.1$	$1.9 \pm 0.2$	$0.5 \pm 0.1$
NIST 1947	$4.9 \pm 0.2$	$0.5 \pm 0.1$	$4.4 \pm 0.2$	$1.0 \pm 0.1$



# Conclusion

- Rapid quantitative extraction technique.
- Integrity of Hg species maintained as minimal interconversion incurred.
- Hg<sup>2+</sup> blank under control.
- No evidence of isotopic fractionation observed.
- Internationally recognised delta value obtained for samples referenced to NIST SRM 3133

## Next step

- Apply the procedure to a range of fish types from different locations.
- More precise collect of fractions less concentrating is required.
- Collected more fractions form a sample digest will increase Hg content of final extract resulting in better data.
- Isotopic analysis of minor  $\text{Hg}^{2+}$  species.

# Acknowledgments

- Panayot Petrov
- Heidi Goenaga-Infante

Funding from -Joint Research Project of the European Metrology Research Programme (ENV51 – MeTra)  
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Questions?